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Synthesis of (1R, 2R)-DPEN-derived triazolium salts and their application in asymmetric intramolecular Stetter reactions[†]

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A series of novel chiral triazolium salts has been synthesized from readily available (1R,2R)-DPEN and found to be efficient for the enantioselective intramolecular Stetter reaction. With 10 mol% of the catalyst, the intramolecular Stetter reaction was realized in excellent yields with up to 97% ee.

N-Heterocyclic carbenes (NHCs) as organocatalysts have received considerable attention in the last two decades.¹⁻² Various kinds of chiral NHCs have been developed to catalyze asymmetric reactions, in which the bicyclic triazolium salts **1–3** are one of the highly efficient classes (Scheme 1). Their bicyclic molecular scaffolds can restrict unfavorable internal rotation around the N–C (substituent) axis, which may enhance the chiral induction. Moreover, the aryl groups on the nitrogen atom may affect both reactivity and selectivity. Because of their structural advantages, these triazolium salts have achieved great success in many asymmetric reactions.³⁻⁵



Scheme 1 Chiral bicyclic triazolium salts.

However, there exist only a limited number of backbone scaffolds for chiral NHCs, and most of them suffer from high cost of the enantiopure starting materials and tedious synthetic procedures. The development of a novel NHC catalyst that can be synthesized from a readily available chiral source and is finetunable with steric and electronic properties is highly desirable.

(1R,2R)-DPEN ((1R,2R)-(+)-diphenyl ethylenediamine) has been used widely as an excellent chiral source for chiral auxiliary and organic synthesis.⁶ Tomioka *et al.* reported the synthesis of dihydroimidazolium salt **4** in 2006, and applied it as organocatalyst in the intramolecular asymmetric Stetter reaction with up to 80% ee.⁷ In 2007, Scheidt and co-workers devised bicyclic triazolium salts **5** derived from the (1S,2R)-(+)-2-amino-1,2diphenyl ethanol backbone and found they were highly efficient in desymmetrization of 1,3-diketones (up to 96% ee)⁸ and addition of homoenolates to nitrones (up to 94% ee).⁹ We therefore envisaged that (1R,2R)-DPEN might be an efficient chiral scaffold for triazolium salts, precursors for NHC catalysts. Compared with the bicyclic triazolium salts **5** derived from (1S,2R)-(+)-2-amino-1, 2-diphenyl ethanol backbone, bicyclic triazolium salts **6** based on (1R,2R)-DPEN scaffold show the advantage of having finetunable R group (Scheme 2).



Scheme 2 Several related N-heterocyclic carbene precursors.

As part of our ongoing endeavors to develop new *N*-heterocyclic carbene catalysts and their applications to novel organocatalytic reactions,¹⁰ in this paper, we report our preliminary results on the synthesis of novel chiral bicyclic triazolium salts from (1R,2R)-DPEN and their application to the asymmetric catalytic intramolecular Stetter reaction.

A series of triazolium salts were easily prepared from the commercially available (1R,2R)-DPEN¹¹ (Scheme 3). With these catalyst precursors in hand, we then examine their catalytic ability in the intramolecular Stetter reaction.



Scheme 3 Synthesis of triazolium salts from (1R, 2R)-DPEN.

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Our studies began with testing different chiral NHCs in the intramolecular Stetter reaction of 9a. As summarized in Table 1, when triazolium salts 6a-6d (20 mol%) were used together with KHDMS (20 mol%), generally appreciable enantioselectivities in the cyclization were afforded (entries 1-4, Table 1). NHC derived from triazolium salt 6c led to the Stetter reaction product 10a in excellent yield and promising asymmetric induction (95% yield and 85% ee, entry 3, Table 1). With 6c, the catalyst loading could be decreased to 10 mol% without significantly affecting the reactivity and only slight erosion of the ee value was observed (entry 5, Table 1). Several conventional bases were found tolerable (entries 1-5, Table 2) and Et₃N was optimal in terms of both yield and ee of the product (entry 6, Table 2). Various solvents such as xylene, CH₂Cl₂, THF, Et₂O, and toluene were tested in this reaction (entries 1-5, Table 3). Finally, reaction in xylene led to an optimal combination of 95% yield and 93% ee.

Under these optimized conditions (that is, **6c**, Et₃N, 0.1 M of substrate in xylene at room temperature), various salicylaldehydederived substrates were tested and the results are summarized in Table 4. Substrates **9b–d** bearing electron-donating groups (5-Me, 5-MeO, 4-MeO, entries 2–4, Table 4) on the salicylaldehydestructure were well tolerated and led to their corresponding chromanone derivatives in good to excellent yields (87–98%) and generally high ees (88–97%). It should be noted that the reaction did not occur when substrate **9e** was used under the standard

Table 1 Screening of different catalysts

| CH O 9a | | 6 (20 mol%) KHMDS (20 mo xylene | 0 1%) 0 10a | COOEt |
|--------------------|----------------------|--|------------------------|----------------------|
| entry ^a | catalyst | time (h) | yield (%) ^b | ee (%) ^c |
| 1 2 3 4 | 6a 6b 6c 6d | 36 36 0.6 36 | 60 43 95 17 | 51 68 85 10 |
| 5 ^d | 6c | 0.6 | 95 | 82 |

^{*a*} Reaction conditions: 0.1 M solution, addition of **9a** to the prior generated catalyst. ^{*b*} Isolated yields. ^{*c*} Determined by chiral HPLC (Daicel Chirapak AD-H). ^{*d*} 10 mol% triazolium salt and 10 mol% KHMDS were used.

Table 2 Screening of bases

| CH O 9a | O COOEt | 6c (10 mol%) base (10 mol%) xylene | | `COOEt |
|--------------------|----------------------------------|--|------------------------|----------------------|
| entry ^a | base | time (h) | yield (%) ^b | ee (%) ^e |
| 1 2 2 | KHMDS NaHMDS | 0.6 1.3 | 95 97 | 82 85 |
| 3 4 5 6 | DBU DIEA Et ₃ N | 48 4 85 16 | 56 90 90 95 | 90 87 93 93 |

^{*a*} Reaction conditions: 0.1 M solution, addition of **9a** to the prior generated catalyst. ^{*b*} Isolated yields. ^{*c*} Determined by chiral HPLC (Daicel Chirapak AD-H).

Table 3Screening of solvents

| CH O 9a | HO COOEt | 6c (10 mol%) Et ₃ N (10 mol%) xylene | + () 0 10a | ^{COOEt} |
|--------------------|------------------|--|------------------------|---------------------|
| entry ^a | solvent | time (h) | yield (%) ^b | ee (%) ^c |
| 1 | xylene | 16 | 95 | 93 |
| 2 | $\dot{C}H_2Cl_2$ | 65 | 82 | 74 |
| 3 | Et_2O | 52 | 79 | 80 |
| 4 | toluene | 22 | 95 | 88 |
| 5 | THF | 42 | 93 | 87 |

^{*a*} Reaction conditions: 0.1 M solution, addition of **9a** to the prior generated catalyst. ^{*b*} Isolated yields. ^{*c*} Determined by chiral HPLC (Daicel Chirapak AD-H).

| R | $\begin{array}{c} CHO \\ X \stackrel{\text{CHO}}{\longrightarrow} COOEt \end{array} \qquad \begin{array}{c} \mathbf{6c} (10 \text{ mc}) \\ \underline{Et_3N (10 \text{ m})} \\ xylene \end{array}$ | 01%) <mark>01%)</mark> → R([| | `COOEt |
|--------------------|--|---------------------------------|------------------------|---------------------|
| | 9 | | 10 | |
| entry ^a | 9 | time (h) | yield (%) ^b | ee (%) ^c |
| 1 | 9a $X = O, n = 1, R = H$ | 16 | 95 | 93 |
| 2 | 9b $X = O, n = 1, R = 5-CH_3$ | 21 | 98 | 95 |
| 3 | 9c $X = O, n = 1, R = 5$ -OCH ₃ | 31 | 95 | 88 |
| 4 | 9d $X = O, n = 1, R = 4$ -OCH ₃ | 26 | 87 | 97 |
| 5 ^d | 9e $X = O$, $n = 1$, $R = 4$ -NEt ₂ | 67 | 56 | 95 |
| 6 | 9f $X = O$, $n = 1$, $R = 3$ -CH ₃ | 8 | 98 | 80 |
| 7 | $9g X = O, n = 1, R = 3-OCH_3$ | 34 | 98 | 81 |
| 8 | 9h $X = O, n = 1, R = 5$ -Cl | 9 | 98 | 78 |
| 9 | 9i $X = O$, $n = 1$, $R = 5$ -Br | 12 | 93 | 78 |
| 10 | 9 j X = S, $n = 1$, R = H | 48 | 70 | 89 |
| 11 | 9k $X = O, n = 1, R = 5 - NO_2$ | 15 | 90 | 0 |
| 12 | 91 $X = O, n = 0, R = H$ | 66 | 59 | 0 |

^{*a*} Reaction conditions: 0.1 M solution, addition of **9** to the prior generated catalyst. ^{*b*} Isolated yields. ^{*c*} Determined by chiral HPLC. ^{*d*} KHMDS was used instead of Et₃N.

conditions. When KHMDS was used instead of Et₃N under the standard conditions, substrate 9e underwent the intramolecular Stetter reaction affording the desired product in 56% yield with 95% ee (entry 5, Table 4). When substrates 9f-g containing substituents adjacent to the linker were used, the reactions ran smoothly with high yield but moderate to good enantioselectivities (entries 6–7, Table 4). The reaction of substrates 9h-i bearing weak electron-drawing groups (5-Cl or 5-Br, entries 8-9, Table 4) led to their desired products in excellent yields but with relatively low ees (78%). The sulfur-tethered substrate 9j gave a moderate yield (70%) with excellent ee (89%) (entry 10, Table 4). When substrate 9k bearing 5-NO₂ group was used, the desired product was obtained in 90% yield but with no enantioselectivity (entry 11, Table 4). Substrate 91 which led to a five-membered ring was also examined and the desired product 10I was obtained also in racemic (entry 12, Table 4). In both cases, they were also known to be challenging substrates in the literature probably due to the ready racemization of the products.^{4a} Notably, in some cases, running the reaction for a prolonged time will cause a certain degree of racemization of the product.

In conclusion, we have developed a series of novel bicyclic triazolium salts from readily available (1R,2R)-DPEN. The NHC catalyst derived from the chiral triazolium salt **6c** and Et₃N is found to be efficient for the enantioselective intramolecular Stetter reaction. In general, excellent yields with up to 97% ee could be obtained for the intramolecular Stetter reaction. Further application of these (1R,2R)-DPEN derived triazolium salts in other asymmetric reactions are currently under way.

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